



WIR SCHAFFEN WISSEN – HEUTE FÜR MORGEN

Anders Kaestner :: Paul Scherrer Institut

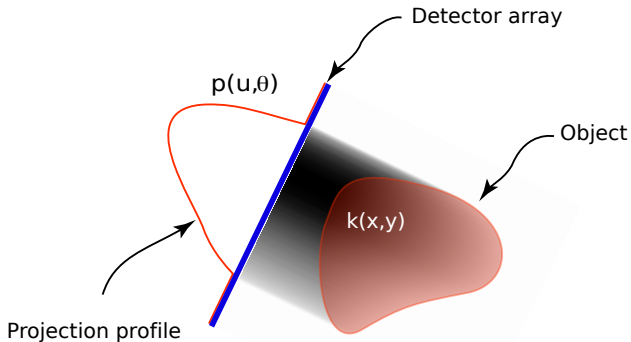
Introduction to Computed Tomography

Part II: Radiography and primitive computed tomography

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- 1 Radiography
 - 2 Tomography
 - 3 Summary

- Understand the image formation process for radiography.
- Motivating the use of computed tomography.

A ray illuminates a semi-transparent medium



A ray penetrating a medium is attenuated according to Beer-Lamberts law The intensity is attenuated in the medium according to

$$I = I_0 e^{-\int_L k(x,y) dl}$$

I - Intensity behind the sample

I_0 - Incident intensity

k - Attenuation coefficient,

μ - Linear attenuation coefficient X-rays

Σ - Macroscopic cross-section for neutrons

L - Line through the sample.

Computing an attenuation image

From Beer-Lamberts law we get

$$p = -\log \left(\frac{r - r_{DC}}{r_{OB} - r_{DC}} \right) = -\log \left(\begin{array}{c|c} \text{[Image 1]} & \text{[Image 2]} \\ \hline \text{[Image 3]} & \text{[Image 4]} \end{array} \right) = \text{[Image 5]}$$

p Normed projection

r Measured radiogram

r_{DC} Dark current image (removes noise floor)

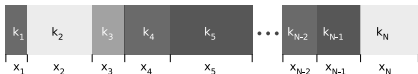
r_{OB} Open beam image, measured I_0

Each pixel represent the line integral $\int_L k(x)dx$ through the sample.

Piecewise constant sample

Few discrete regions

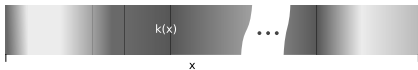
$$I = I_0 e^{-\sum_{i=1}^N k_i x_i}$$



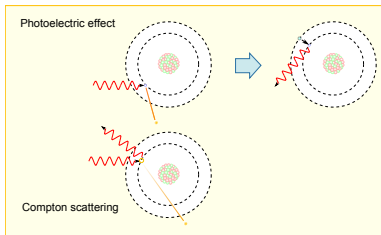
Continuous samples

Let $x_i = \Delta x$ and $\Delta x \rightarrow 0$

$$I = I_0 e^{-\int_L k(x) dx}$$

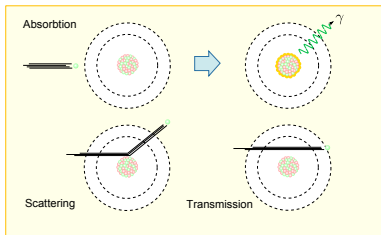


X-rays



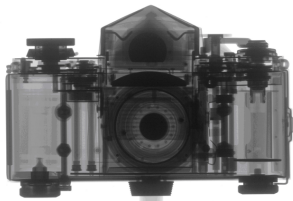
- Electromagnetic radiation.
- Interaction with the electron shells.

Neutrons



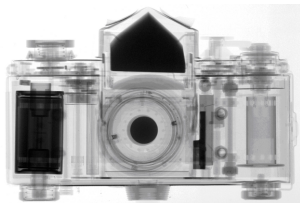
- Neutral particle beam.
- Interaction with the nucleus.

X-rays at 150keV



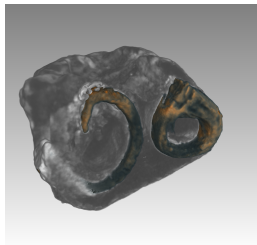
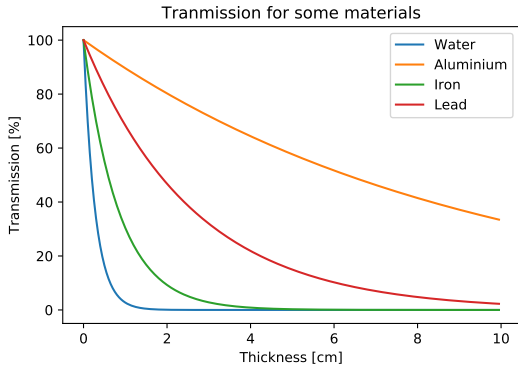
| Group → | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 |
|-------------|-------------|-------------|-------------|-------------|-------------|------------|-------------|-------------|-------------|-------------|-------------|-------------|-------------|-------------|-------------|-------------|-------------|-------------|
| 1 | H 1.02 | | | | | | | | | | | | | | | | | He 0.02 |
| 2 | Li 0.08 | Be 0.22 | | | | | | | | | | | B 0.28 | C 0.27 | N 0.11 | O 0.16 | F 0.14 | Ne 0.17 |
| 3 | Na 0.13 | Mg 0.24 | | | | | | | | | | | Al 0.38 | Si 0.33 | P 0.25 | S 0.30 | Cl 0.23 | Ar 0.15 |
| 4 | K 0.14 | Ca 0.28 | Sc 0.48 | Ti 0.73 | V 1.04 | Cr 1.29 | Mn 1.52 | Fe 1.57 | Ni 1.78 | Cu 1.98 | Zn 1.84 | Ga 1.45 | Ge 1.33 | As 1.28 | Se 1.23 | Br 0.90 | Kr 0.73 | |
| 5 | Rb 0.47 | Sr 0.86 | Y 1.61 | Zr 2.47 | Nb 3.43 | Mo 4.29 | Tc 5.59 | Ru 5.71 | Rh 6.15 | Pd 5.67 | Cd 4.84 | In 4.31 | Sn 3.98 | Sb 4.23 | Te 4.65 | I 3.45 | Xe 2.53 | |
| 6 | Cs 1.47 | Ba 2.73 | | Hf 19.76 | Ta 25.47 | W 35.48 | Re 34.47 | Os 27.80 | Ir 30.01 | Pt 30.61 | Au 35.94 | Hg 25.88 | Tl 22.23 | Pb 25.81 | Bi 20.28 | Po 20.52 | At 19.77 | Rn 17.77 |
| 7 | Fr 11.68 | Ra 11.68 | | Rf - | Db - | Sg - | Bh - | Hs - | Mt - | Ds - | Rg - | Uub - | Uuq - | Uup - | Uuh - | Uus - | Uuo - | |
| Lanthanides | | | | | | | | | | | | | | | | | | |
| | La 5.04 | Ce 2.79 | Pr 4.25 | Nd 6.45 | Pm 7.33 | Sm 7.68 | Eu 5.89 | Gd 6.69 | Tb 9.41 | Dy 10.17 | Ho 10.17 | Er 11.30 | Tm 12.40 | Yb 9.32 | Lu 14.07 | | | |
| Actinides | | | | | | | | | | | | | | | | | | |
| | Ac 24.47 | Th 38.95 | Pa 39.62 | U 49.68 | Np - | Pu - | Am - | Cm - | Bk - | Cf - | Es - | Fm - | Md - | No - | Lr - | | | |

Thermal neutrons



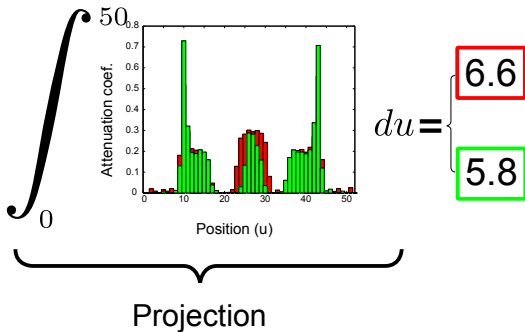
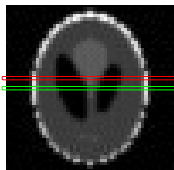
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|-------------|------------|------------|------------|------------|------------|-------------|--------------|-------------|--------------|-------------|------------|-------------|------------|------------|------------|------------|------------|------------|
| 1 | H 3.34 | | | | | | | | | | | | | | | | | He 0.02 |
| 2 | Li 0.30 | Be 0.75 | | | | | | | | | | | | | | | | Ne 0.10 |
| 3 | Na 0.35 | Mg 0.35 | | | | | | | | | | | | | | | | Ar 0.53 |
| 4 | K 0.06 | Ca 0.08 | Sc 2.02 | Ti 0.60 | V 0.82 | Cr 0.72 | Mn 0.94 | Fe 1.21 | Ni 1.19 | Cu 1.02 | Zn 2.05 | Cd 1.07 | Zn 0.35 | Ga 0.48 | Ge 0.47 | As 0.67 | Se 0.73 | Kr 0.81 |
| 5 | Rb 0.08 | Sr 0.14 | Y 0.27 | Zr 0.29 | Nb 0.40 | Mo 0.52 | Tc 1.76 | Ru 0.58 | Rh 16.48 | Pd 0.76 | Ag 4.04 | Cd 1.51 | Sn 1.08 | Sb 0.21 | Se 0.30 | Te 0.25 | I 0.23 | Xe 0.43 |
| 6 | Cs 0.29 | Ba 0.07 | | Hf 0.99 | Ta 1.68 | W 1.63 | Re 0.84 | Os 0.28 | Ir 39.46 | Pt 1.63 | Au 0.21 | Hg 16.21 | Tl 0.47 | Pb 0.38 | Bi 0.27 | Po - | At - | Rn - |
| 7 | Fr - | Ra 0.38 | | Rf - | Db - | Sg - | Bh - | Hs - | Mt - | Ds - | Rg - | Uub - | Uuq - | Uup - | Uuh - | Uus - | Uuo - | |
| Lanthanides | | | | | | | | | | | | | | | | | | |
| | La 0.52 | Ce 0.14 | Pr 0.41 | Nd 1.83 | Pm - | Sm 0.72 | Eu 171.47 | Gd 44.58 | Tb 147.83 | Dy 10.83 | Ho 2.25 | Er 5.48 | Tm 3.53 | Yb 1.46 | Lu 2.75 | | | |
| Actinides | | | | | | | | | | | | | | | | | | |
| | Ac - | Th 0.29 | Pa 0.46 | U 0.82 | Np 9.81 | Pu 15.20 | Am 2.98 | Cm - | Bk - | Cf - | Es - | Fm - | Md - | No - | Lr - | | | |

Some attenuation examples for neutrons



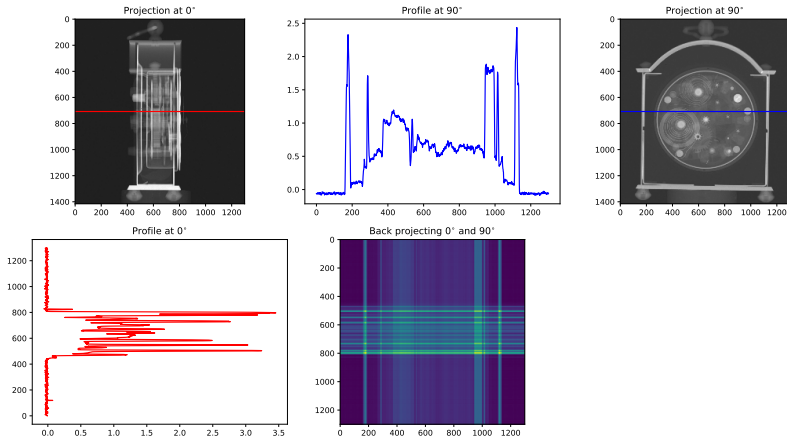
Neutron tomography of fist-sized lead canon ball from the battle of Bosworth (1485AD)

Limitation of the radiography



- Great local changes buried in the sum of bulk
- Depth position can't be determined

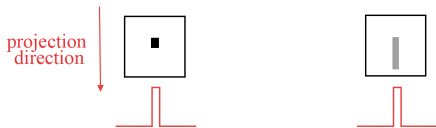
Use two projections at 90° to get depth information



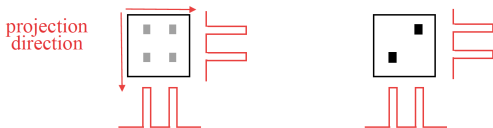
- Provides some depth information
- Still a lot of guessing

The solution is not unique

Single projection \rightarrow several solutions



Two projections \rightarrow several solutions



A unique solution would exist only for an infinite number of noiseless continuous projections

What is tomography?

- A method to capture three-dimensional images.
- An indirect method using projections (radiograms) to reconstruct the inner structure of a sample.
- Free translation is slice imaging
from Greek:

Tomos – 'a section' or 'a cutting'

Graph – write

- 1917 **Radon** developed the foundation for the inversion required by tomography.
- 1956 **Bracewell** the relationships between Fourier transform and Radon transform.
- 1963 First applications to medical tomography.
Kuhl obtained first backprojection.
Cormack applied Radon's results to radiograms.
- 1970 Publication of the first CT image.
- 1970-1973 **Cormack & Hounsfield** first CT scanner.
- 1979 **Cormack & Hounsfield** the Nobel prize in Medicine.



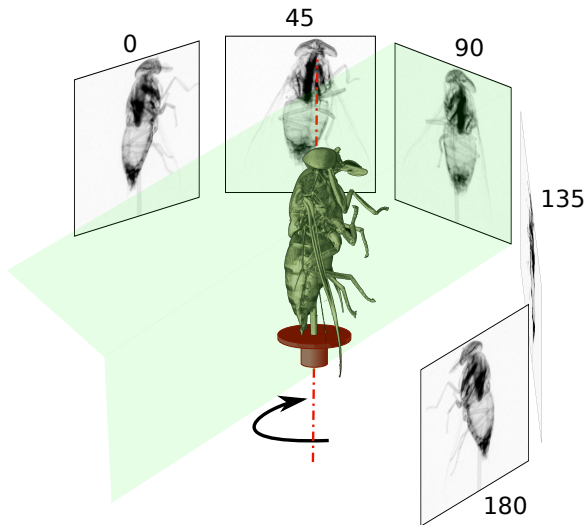
J. Radon (1887–1956)

R. Bracewell
(1921–2007)

D Kuhn (1929–2017)

A. Cormack
(1924–1998)Sir G.N. Hounsfield
(1919–2004)

Inspecting the sample from different views



Observations

$$2 \quad 3 \quad \rightarrow \quad 5$$

$$1 \quad 4 \quad \rightarrow \quad 5$$

$$\downarrow \quad \downarrow$$

$$3 \quad 7$$

Equation system

$$a_{11}x_1 + a_{12}x_2 = y_1$$

$$a_{21}x_3 + a_{22}x_4 = y_2$$

$$a_{11}x_1 + a_{21}x_3 = y_3$$

$$a_{12}x_2 + a_{22}x_4 = y_4$$

$$\vdots$$

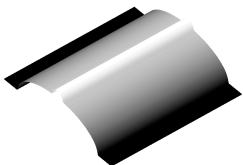
$$\Rightarrow Ax = y$$

Solve the equation system for x

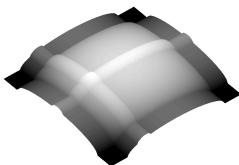
Many equations, sparse matrix A , no unique solution...

A first attempt to reconstruction: Back-projection

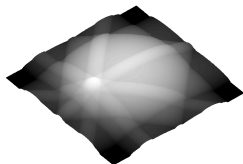
1 projections



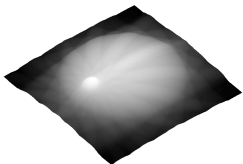
2 projections



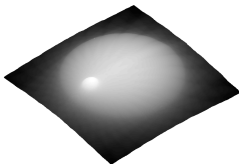
4 projections



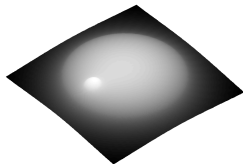
8 projections



16 projections



32 projections



The solution is too smooth. . . something is missing!!!

- Radiography to inspect interiors of opaque objects.
- Different sources and the attenuation law.
- First motivating steps towards computed tomography.



Sears, V. (1992).

Neutron scattering lengths and cross sections.

Neutron News, 3(3):26–37.